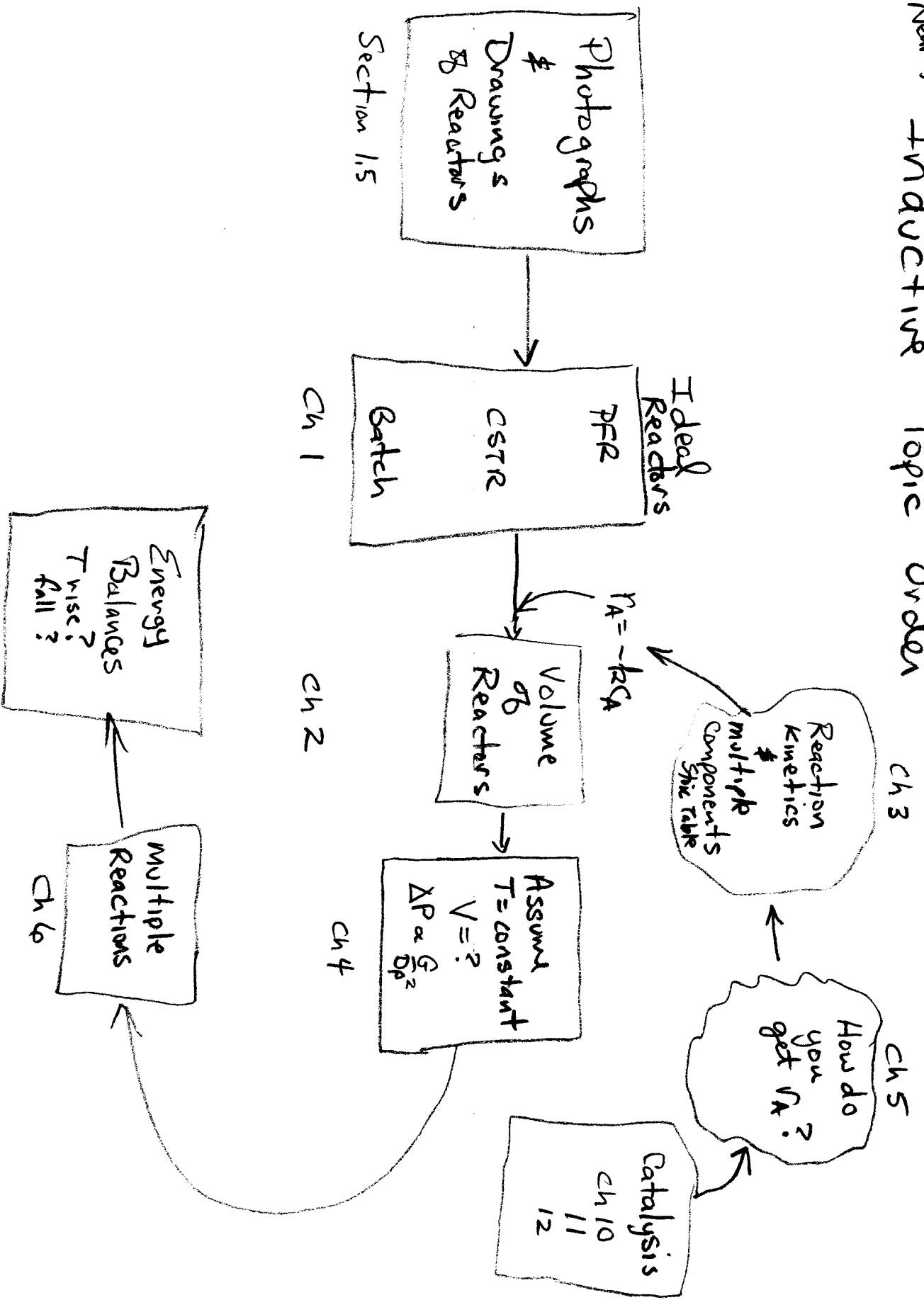


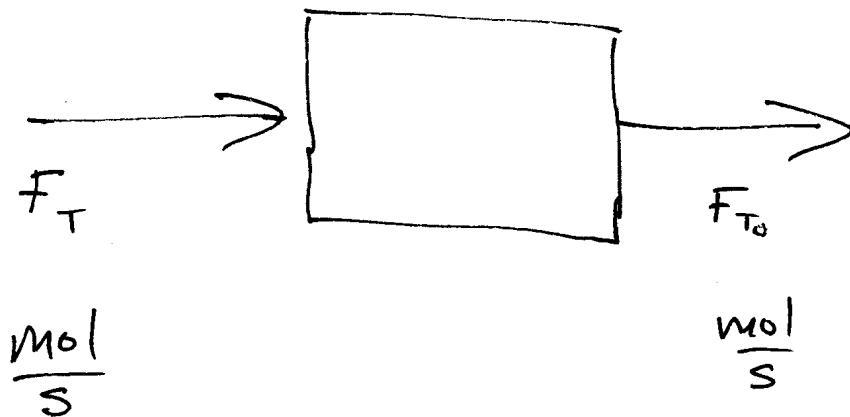
"Newly" Inductive Topic Order



Objectives

1. Give the definition of conversion for a flow system
2. Construct a stoichiometric table.

In CRE we use moles

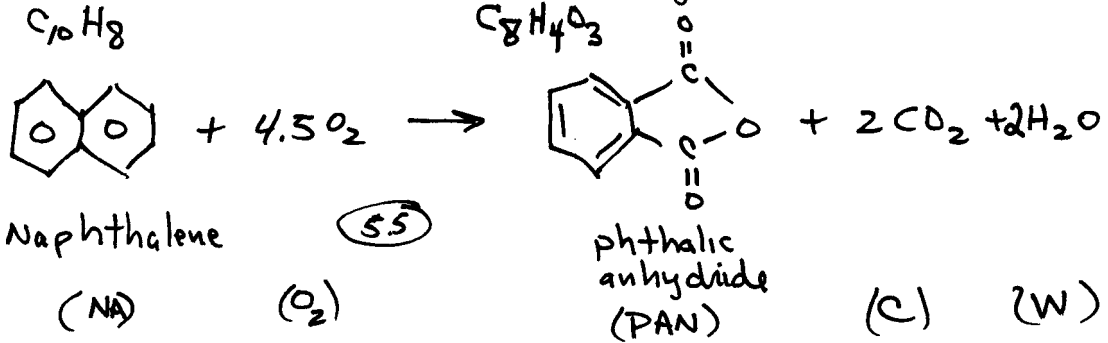


Help for Reactor Design Project

memo 1

See Section 3.6 in Fogler

Formation of phthalic anhydride (GAS PHASE)



Question: what do you expect for an overall change in total # moles +0.5 moles

Define conversion of NA

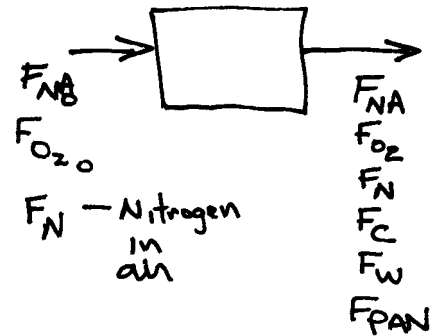
$$X_{NA} \equiv \frac{F_{NA0} - F_{NA}}{F_{NA0}}$$

F_{NA0} [] has units $\frac{\text{mol NA fed to Reactor}}{s}$

Rearranging

$$F_{NA} = F_{NA0} - F_{NA0} X_{NA}$$

↑ definition
 ↑ initial ↓ change



Species Initial Molar change Final

NA	F_{NA0}	$- F_{NA0} X_{NA}$	$F_{NA} = F_{NA0} - F_{NA0} X_{NA}$
O ₂	F_{O20}	$F_{NA0} X_{NA}$	
N	F_{N0}	$F_{NA0} X_{NA}$	
C	0	$F_{NA0} X_{NA}$	
W	0	$F_{NA0} X_{NA}$	
PAN	0	$F_{NA0} X_{NA}$	

See next page

How do you write F_{O2} in terms of X_{NA} ?

$$F_{O2} = F_{O20} - \frac{4.5 \text{ mol } O_2}{1 \text{ mol NA}} F_{NA0} X_{NA}$$

write next

<u>Species</u>	<u>Initial</u>	<u>Molar Change</u>	<u>Final</u>
NA	F_{NA_0}	$- F_{NA_0} X_{NA}$	$F_{NA} = F_{NA_0} X_{NA}$
O ₂	$F_{O_2_0}$	$- \frac{4.5 \text{ mol O}_2 \Delta}{1 \text{ mol NA} \Delta} F_{NA_0} X_{NA}$ $\frac{\text{mol NA}_0 (\text{mol NA} \Delta)}{\text{mol NA}_0}$ $\text{mol O}_2 \Delta$	$F_{O_2} = F_{O_2_0} - 4.5 F_{NA_0} X_{NA}$
N	F_{N_0}	0	$F_N = F_{N_0}$
C	F_{C_0}	$+ \frac{2 \text{ mol CO}_2 \Delta}{1 \text{ mol NA} \Delta} F_{NA_0} X_{NA}$ $\frac{\text{mol NA}_0 (\text{mol NA} \Delta)}{\text{mol NA}_0}$	$F_C = F_{C_0} + 2 F_{NA_0} X_{NA}$
W	F_{W_0}	$+ \frac{1 \text{ mol W} \Delta}{1 \text{ mol NA} \Delta} F_{NA_0} X_{NA}$	$F_W = F_{W_0} + F_{NA_0} X_{NA}$
PAN	F_{PAN_0}	$+ \frac{1 \text{ mol PAN} \Delta}{1 \text{ mol NA} \Delta} F_{NA_0} X_{NA}$	$F_{PAN} = F_{PAN_0} + F_{NA_0} X_{NA}$

Examine chart

<u>Species</u>	<u>Initial</u>	<u>Molar change</u>	<u>Final</u>
		$() F_{NA_0} X_{NA}$	
		$() F_{NA_0} X_{NA}$	
		\vdots	
		$() F_{NA_0} X_{NA}$	

This column has $F_{NA_0} X_{NA}$

Examine totals

Initial:

$$F_{T_0} = \sum F_{i_0} = F_{NA_0} + F_{O_{2_0}} + F_{N_0} + 0 + 0 + 0$$

what is F_{N_0} ?

$$\begin{aligned} &0.21 O_2 \\ &\sim 0.79 N_2 \end{aligned}$$

$$F_{N_0} = \frac{0.79 \text{ mol } N_2}{0.21 \text{ mol } O_2} F_{O_{2_0}}$$

Final

$$F_T = \sum F_i$$

$$F_T = F_{T_0} + \left[-4.5 F_{NA_0} \chi_{NA} - F_{NA_0} \chi_{NA} + 2 F_{NA_0} \chi_{NA} + F_{NA_0} \chi_{NA} + F_{NA_0} \chi_{NA} \right]$$

$$F_T = F_{T_0} + F_{NA_0} \chi_{NA} \left[\underbrace{-1 - 4.5 + 2 + 2 + 1}_{-0.5} \right] = \left(\frac{-5.5 + 5}{-0.5} \right)$$

$$F_T = F_{T_0} - 0.5 F_{NA_0} \chi_A$$